CH₄ Flux in an Alas Ecosystem Formed by Forest Disturbance near Yakutsk, Eastern Siberia, Russia

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Alas is a round area of grassland with a pond at the center, formed by subsidence associated with permafrost thawing in Taiga forests in eastern Siberia. Some Alases have a Pingo, which is a round mound formed by the refreezing of the Alas pond. To characterize the relationship between the CH₄ dynamics and soil properties in a forest-Alas ecosystem, we investigated 2 line transects in an Alas near Yakutsk: one transect from the forest to the pond (300 m), and the other from the top of the Pingo to the pond (26 m). Both soil pH and EC in the topsoil were higher in the grassland and the Pingo (5.8-7.1 for pH and 7-141 mS m⁻¹ for EC) than in the forest (4.3 for pH and 1-17 mS m⁻¹ for EC). The organic carbon contents of the soils were also higher in the grassland and the Pingo than in the forest. CH₄ uptake in the forest soils ranged from -13 to $-17 \mu g \text{ C m}^{-2} \text{ h}^{-1}$. On the other hand, CH_4 emission in the grassland increased from -8 to 3.8×10^3 µg C m⁻² h⁻¹ with the approach to the pond. Large CH₄ emissions of 0.6 to 7.0×10^3 µg C m⁻² h⁻¹ were observed at the edge of the pond. The values of the CH₄ fluxes were similar to those previously reported -25 to -5 $\mu g \ C \ m^{-2} \ h^{-1}$ for forests and 0.1 to $5.2 \times 10^3 \ \mu g \ C \ m^{-2} \ h^{-1}$ for wetlands in Alaska and northern Europe. CH₄ fluxes were strongly correlated with the soil moisture content, but not with the soil temperature. No CH₄ flux was observed in the Pingo soils, probably due to the large accumulation of salts. These results suggest that Alas formation with forest disturbance increases CH₄ production and decreases CH₄ consumption due to the accumulation of salts and the increase in the soil moisture and organic carbon content in the Alas soil.

Key Words: Alas, forest fire, methane, permafrost, Taiga.

CH₄ is an important greenhouse gas with a contribution of about 20% to the radiative forcing of the global climate (IPCC 2001). Total CH₄ emission is estimated of 598 Tg year⁻¹ (IPCC 2001). Major sources of atmospheric CH₄ include the production and use of energy, biomass burning, rice fields, natural wetland systems, landfills, and enteric fermentation in animals and termites. The major sinks are considered to be the atmosphere and soils: 470 Tg year⁻¹ is estimated to be consumed by the reaction of CH₄ with hydroxyl radicals in the atmosphere, while forest soils consume 30 Tg year⁻¹ (IPCC 2001). However, little or no information is available about the CH₄ dynamics in the vast low-rainfall coniferous forests of eastern Siberia (Smith et al. 2000).

Eastern Siberia is characterized by a low precipitation (209 mm as annual average, Robert 1997) and low temperature (-10.4°C as annual average, Robert 1997), presence of permafrost (Desyatkin 1993), and occasion-

al wildfires (Goldammer and Furyaev 1996). Larch species, Larix gmelinii and Larix cajander, are predominant because of their high regeneration ability after wildfire (Abaimov 1995; Schulze et al. 1995; Kajimoto et al. 1999). However, some fires are strong enough to completely destroy the forest ecosystems due to subsidence after permafrost thawing. Isaev et al. (2002) using remote sensing data and forest inventory data showed that, 22% of a studied forest area of 24,700 ha in eastern Siberia was damaged by high-intensity ground fires which completely destroyed forest ecosystems. They stared their 70-80% of the fires were caused by human activity with the remainder caused by natural events such as lightning (Goldammer and Stocks 2000). After severe forest fires, Alases are formed (Desyatkin 1993). Alases are thermokarsts and water supplied from the thawing permafrost flows laterally into the Alas, forming ponds with diameters ranging from several tens of meters to 100 m. Some Alases have a Pingo which is a small hill (10-20 m in diameter, 10-20 m in height) formed by the refreezing of water in the pond (Fitzpatrick 1983). CH₄ flux from soils is affected by the soil temperature and moisture (Castro et al. 1994; Whalen and Reeburgh 1996; Primè and Christensen 1997; Bowden et al. 1998). Microbiological processes regulate the production and consumption of CH₄ in soil (Mer and Roger 2001). In general, forest soils take up CH₄ (Smith et al. 2000), whereas wetlands emit CH₄ (Mer and Roger 2001). As the land-cover modifies often changes the thermo-hydrological conditions of soil, forest disturbance is generally considered to decrease the CH₄ consumption activity of soil (Dobbie et al. 1996). The disappearance of forest may decrease CH4 uptake in the Taiga ecosystem, and the appearance of ponds may increase CH₄ emission.

The purpose of this study was to characterize the relationship between the CH₄ dynamics and soil properties in a forest-Alas ecosystem near Yakutsk, eastern Siberia.

MATERIALS AND METHODS

Site description. This study was conducted in an Alas and adjacent forest in Neleger, located 30 km north-northwest from the city of Yakutsk (lat $62^{\circ}05'$ N, long $129^{\circ}45'$ E), Russia (Fig. 1). The Alas is about 500 m \times 1,000 m, in size with a pond 100 m in diameter and 1 m deep. All the measurements were carried out in July 2000, and June and July 2001.

The forest consists mainly of Larch (*Larix gmelinii*) trees more than 200 years old, with *Vaccinium vitisidaea* predominatily on the forest floor. The grassland of the Alas consists predominantly of *Elytriga repens* near the forest and *Carex resucata* near the pond. *Potentilla anserina* predominates on the Pingo.

Measurements.

Set up of line transects: The location and topography of the 2 line transects are shown in Figs. 1, 2, and 4. One transect (300 m) was set up from the forest to the pond through the grassland (the FG-line), and the other transect (26 m) was set up from the top of the Pingo to the pond (the P-line). The location of the grassland was 2 m lower than that of the forest and there was a steep slope at the edge of the forest (Fig. 2). One of the measuring points was set up in the forest (A) and the other points were set up through the grassland at every 10 to 20 m from the edge of the forest to the pond (B-E in 2000, B-F in 2001) (Figs. 1 and 2). The water depth at E in 2000 or F in 2001 was 10 cm. In 2001, E was in the grassland because the area of the pond in 2001 had decreased compared to the area in 2000. On the P-line, 14 measurement points were set up at every 2 m inter-

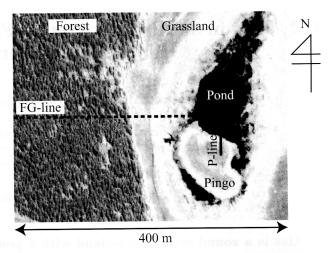


Fig. 1. IKONOS image of Neleger on July 11, 2001 provided by (C) Japan Space Imaging Corporation.

vals from the top of the Pingo to the edge of the pond $(P_0, P_2, P_4, ..., P_{22}, P_{24}, P_{26};$ numbers represent the distance in m from the top of the Pingo). The top of the Pingo was located on 5 m above the pond surface (Fig. 4).

Soil environment: Examination of the soil profile

and soil sampling were conducted in the forest (near A), in the grassland (between B and C, and near D), and on the top of the Pingo (Fig. 1). Intact 100-mL cores (5 cm diameter, 3 replications) and bulk samples (PVC bag: about 500 mL) were collected from each horizon of each soil profile. Using the intact cores, the bulk density was determined gravimetrically by drying to a constant weight at 105°C overnight. Bulk samples were air-dried for more than 3 weeks in the laboratory, and then sieved through a 2-mm-mesh sieve for use for chemical analysis.

Soil pH was determined with a glass electrode pH meter in a supernatant suspension of 1:2.5 (O horizon, 1:20) soil: deionized water mixture. EC was determined with a EC meter in a 1:5 soil: deionized water mixture. The EC value was modified by multiplying by 5 to approximate the EC value for a water-saturated soil paste. After the soil samples were air-dried and ground, total carbon and nitrogen contents were determined by using a C/N analyzer (NC-1000, Sumika Chemical Analysis Service, Ltd., Osaka, Japan). Carbonate-carbon content was analyzed by measuring the amount of CO₂ produced from 0.3 to 0.8 g of soil samples to which 15 mL of 2 M HCl solution was added in an Erlenmeyer flask. Organic carbon content was calculated by subtracting the carbonate-carbon content from the total carbon content (Loeppert and Suarez 1996; Sawamoto et

Soil temperature and moisture: Soil temperature and soil moisture were measured at each point on the line

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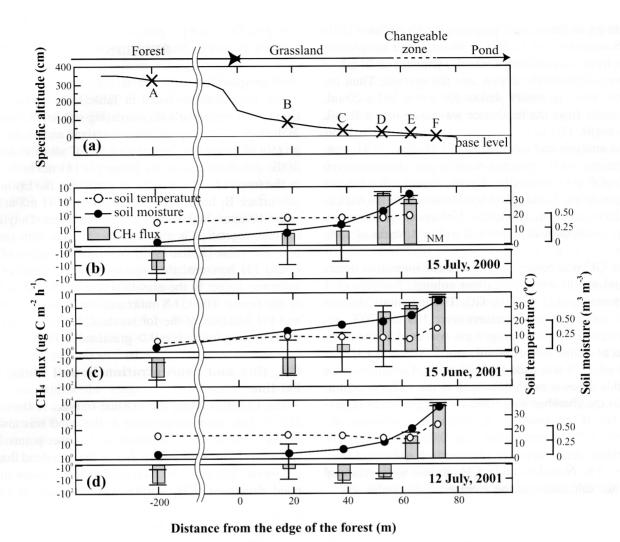


Fig. 2. Topography and CH₄ fluxes along FG-line. (a) Topography. Base level of specific altitude is located on the surface of the pond. (b), (c), and (d) CH₄ fluxes with soil temperature and moisture in July 2000; June 2001; and July 2001; respectively. CH₄ fluxes are indicated on a log scale and vertical bars denote maximum and minimum values of CH₄ flux. NM, not measured.

transects $(A-F, P_0-P_{26})$. Soil temperature was measured with a digital thermometer at a depth of 4 cm. Soil moisture was measured as volumetric water content with a FDR (ML2 Theta Probe, Delta-T Devices, Co., Cambridge, UK) at a depth of 0 to 7 cm. All the measurements were taken with 5 replications.

 ${
m CH_4}$ flux: ${
m CH_4}$ flux measurements were carried out by applying a closed chamber technique. To transport the chambers into the field more conveniently, the chambers were constructed so as to allow 6 chambers of slightly differing sizes to fit into the largest of them in the same way as a Russian matryoshka doll. That is, the height of each of the 6 chambers was similar (25 cm), but the diamaeter was different (18.5, 19, 19.5, 20, 20.5, and 21.0 cm). The ${
m CH_4}$ fluxes were measured at A–F on the FG-line (3 to 6 replications) and at ${
m P_0-P_{26}}$ on the P-line (no replication). ${
m CH_4}$ fluxes on the P-line were measured only in June 2001. On the day before the measurement, chambers were installed at 3 cm depth into the

soil. At 0, 10, 20, and 40 min after the chamber was set up, a 20-mL gas sample was taken into a 10-mL glass bottle vacuum-sealed with a butyl rubber stopper and a plastic cap.

CH₄ concentration in the soil: To collect gases in the soil, stainless steel pipes 9-mm in diameter were installed at depths of 5, 10, 15, 20, 30, and 50 cm (3 pipes at each depth). After the pipes were set up, 50 mL of air was exhausted from each pipe. The pipes were then sealed by using a 3-way cock and kept overnight to allow gas concentrations in the pipes to equilibrate with the soil air. From each of the pipes installed at each depth, air samples (50 mL) were taken into a Tedlar bag. At the same time, air on the soil surface was taken into a Tedlar bag as a sample of soil air at a depth of 0 cm. Within the same day, 20-mL air samples from each of the Tedlar bags were transferred to 10-mL glass bottles.

Sometimes we could get only water from the pipes. In that case, dissolved CH₄ concentrations were measured,

according to the method proposed by McAullife (1971) and Sawamoto et al. (2002). A 30-mL water sample was taken from the pipe with a 60-mL syringe, and 30 mL of air was immediately drawn into the syringe. Then the syringe was vigorously shaken for 3 min and a 20-mL air sample from the headspace was taken into a 10-mL glass bottle.

Gas analysis and calculation of CH₄ flux: CH₄ concentrations were analyzed with a gas chromatograph (GC) (GC-8A; Shimadzu, Kyoto, Japan) after the gas samples in the 10-mL glass bottles were brought back to the laboratory of Hokkaido University in Sapporo, Japan. Analysis was performed within 3 weeks of sampling.

The GC was equipped with a flame ionization detector and a 2-m activated carbon column. Samples of 1 mL were injected into the GC. The injection/detection and column oven temperatures were 130 and 70°C, respectively. Ultrapure nitrogen gas was used as the carrier gas at a flow rate of 60 mL min⁻¹. Standard CH₄ gas (1.95 μ L L⁻¹) was used for calibration. Gas fluxes were calculated based on the changes in the CH₄ concentration in the chamber with time, by using a linear regression law (Christensen et al. 1995). The minimum detectable CH₄ concentration was 0.02 μ L L⁻¹ and the minimum detectable flux (*F*) was -8 < F (μ g C m⁻² h⁻¹) < +8. Non-detectable CH₄ fluxes were removed from the calculation of the average.

RESULTS

Soil properties

Soil properties are listed in Table 1. At all the sites, the soil pH increased with increasing soil depth. Soil pH in the topsoil was higher in the grassland and on the Pingo (5.8-7.1) than in the forest (4.3). EC was also higher in the grassland and on the Pingo $(7-141 \text{ mS m}^{-1})$ than in the forest $(1-17 \text{ mS m}^{-1})$. In particular, the EC in the subsurface B₁ horizon of the Pingo was 141 mS m⁻¹, a value 10 times higher than that in the forest. Only salt-tolerant vegetation is able to grow in soils with such a high EC value (Keren 1999). With the exception of the L and FH horizons, the organic carbon content of the soils was higher in the grassland and on the Pingo than in the forest. The C/N ratio was the highest in the L and FH horizons of the forest; the C/N ratios were in the following order: forest > grassland > Pingo.

CH₄ flux and concentration in soil along the FG-line

The CH₄ flux along the FG-line transect is shown in Fig. 2. The water temperature in the pond was always higher than the soil temperature at the other points. The soil temperature was also higher in the grassland than in the forest. Soil moisture increased as the distance to the pond decreased. CH₄ uptake was observed in forest

Table 1. Properties of forest and grassland soils in Neleger.

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Layer	Depth (cm)	Soil texture	Bulk density (Mg m ⁻³)	pН	EC (mS m^{-1})	Org-C (%)	T-C (%)	T-N (%)	C/N
Forest									
L	-10	SiCL		4.3	17	50.3	50.3	0.58	86
FH	-7	SiCL		4.5	7	52.0	52.0	1.43	36
E1	2	SiCL	1.12	4.8	1	1.8	1.8	0.04	47
E2	25	SiCL	1.51	6.3	1	0.6	0.6	0.03	22
B2 _h	56	SiCL	1.51	8.7	9	0.5	0.5	0.03	17
В3	73	SiCL	1.50	9.2	10	0.4	0.9	0.03	35
C	108 +	SiCL	1.52	9.1	11 000	0.5	1.0	0.03	40
Grassland									
A1	12	SiC	0.72	6.7	7	10.4	10.4	0.98	11
A2	19	SiC	1.52	8.7	12	1.2	2.0	0.06	35
$B2_{ir}$	31.5	SiC	. 1.55	9.0	18	0.9	1.8	0.04	52
B2 _h	49	SiC	1.60	8.7	29	0.2	0.9	0.03	32
B3 _g	78	SiC	1.53	8.5	31	0.2	0.8	0.03	31
$C_g^{}$	120+	SiC	1.50	8.3	26	0.9	1.1	0.05	23
Grassland									
A	20	SiC	1.08	7.1	36	13.3	13.7	1.06	13
AB	28	SiC	1.36	8.8	21	1.7	1.98	0.13	15
$B1_g$	50	SiC	1.63	8.8	36	0.7	0.74	0.05	15
$B2_g^g$	130+	SiC	1.60	8.4	42	0.5	1.12	0.04	28
Pingo									
A	7	SiC	0.95	5.8	12	4.4	4.41	0.33	13
B1	36	SiC	0.84	4.5	141	5.1	5.07	0.41	12
B2	130+	SiC	1.14	8.2	19	1.2	1.66	0.05	33

soils; the CH₄ flux ranged from -13 to $-17~\mu g$ C m⁻² h⁻¹. CH₄ emission in the grassland increased with the approach to the pond—from $-8~\mu g$ C m⁻² h⁻¹ at B to $3.8\times10^3~\mu g$ C m⁻² h⁻¹ at E. Large CH₄ emissions of 0.6 to $7.0\times10^3~\mu g$ C m⁻² h⁻¹ were recorded at E and F. The values of the CH₄ uptake in the forest were similar in these three measurements. On the other hand, CH₄ emission in the grassland significantly changed; that is, the CH₄ flux was higher in July 2000 than in July 2001. Notably, CH₄ uptake was observed at B, C, and D in July 2001.

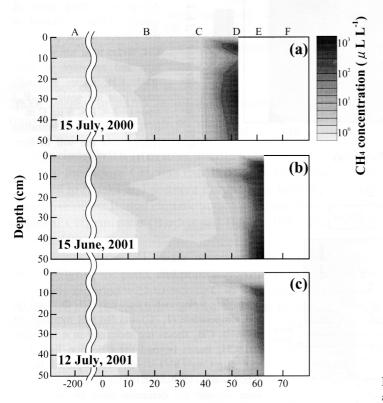
CH₄ concentration in the forest soil decreased with increasing soil depth (Fig. 3). The CH₄ concentration in the grassland soil (B and C), where a very small CH₄ flux was observed, increased slightly with increasing soil depth. However, the distribution of the CH₄ concentration in the soil at C and D in July 2000 was similar to that at D and E in June and July 2001. The CH₄ concentration in the soil at E was 100 times higher than that in the atmosphere (about 2 μ L L⁻¹). The area of soil with the highest concentration of CH₄ (>10 μ L L⁻¹)—i.e., the area within the range of 10 m surrounding the pond—decreased with a 10-m decrease in the pond radius from 2000 to 2001. It was also observed that the CH₄ emission decreased as the area of the pond decreased (Fig. 2).

$\mathrm{CH_4}$ flux and concentration in soil along the P-line

The CH₄ flux and concentration in soil on the P-line in June 2001 are shown in Fig. 4. Soil temperature decreased from the top of the Pingo (P_0) to P_6 , but it increased from P₈ to P₁₂ and decreased again toward the pond. The water temperature was 5 to 10°C higher than the soil temperature. Soil moisture decreased with increasing soil temperature. Because the P-line was drawn from south to north (Fig. 1), the top of the Pingo and sites P₈-P₁₂ were well exposed to sunshine. Therefore, the soil temperature and moisture changed depending on the topography. At sites P₀ to P₁₆, CH₄ fluxes were below the detection limit $[-8 < F \, (\mu g \, C \, m^{-2} \, h^{-1}) < +8]$ (no detectable emission or uptake), while from P₁₈ to the pond, large CH₄ emissions ranging from 21 to 189 μg C m⁻² h⁻¹ were observed, similar to those at D and E on the FG-line. CH₄ concentrations in the soil closely reflected CH₄ fluxes; that is, there was no difference between the CH⁴ concentration in the soil and atmosphere at P_0-P_{16} , where CH_4 fluxes were below the detection limit, whereas from P₁₄ to the pond, the CH₄ concentration in soil was higher than that in the atmosphere.

Relationship between CH₄ flux and soil temperature or moisture

The relationship between the CH₄ flux and soil tem-



Distance from the edge of the forest (m)

Fig. 3. CH₄ concentration (μ L L⁻¹) in soil or water along the FG-line in July 2000 (a); June 2001 (b); and July 2001 (c); respectively.

perature or soil moisture is shown in Fig. 5. The CH₄ fluxes outside the detection limit $[-8 < F \text{ (}\mu\text{g C m}^{-2} \text{ h}^{-1}) < +8]$ are depicted in the "trace zone"; there were many such fluxes in this study. There was no relationship between the CH₄ flux and soil temperature (Fig. 5a). On the other hand, there was a close relationship between the CH₄ flux and soil moisture (Fig. 5b). CH₄ uptake occurred in the forest where the soil moisture content was low, and CH₄ emissions occurred in the pond and the edge of the pond where the soil moisture content was high; both uptake and emission of CH₄

occurred in the grassland where the soil moisture conditions varied. The CH₄ flux on the Pingo showed a similar tendency to that in the grassland. However, at a low soil moisture, CH₄ uptake was observed in the grassland, but not on the Pingo. Soil moisture contents at A, B, and C in July 2000 were lower than in June 2001 and higher than in July 2001, but there was no conspicuous difference in the CH₄ flux among them (Fig. 6a). On the other hand, CH₄ emissions at D and E decreased (Fig. 6b) with decreasing soil moisture from July 2000 to July 2001. At D in particular, CH₄ emission was observed

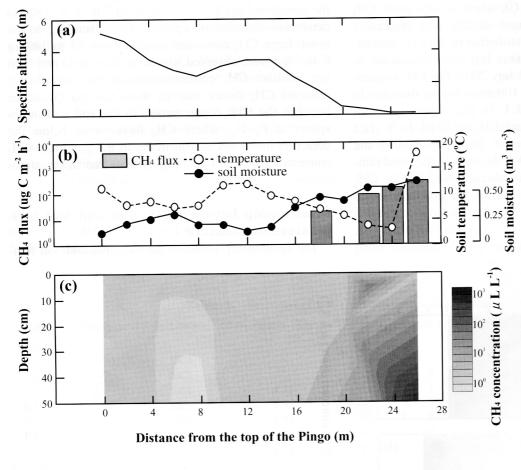


Fig. 4. Topography, CH_4 flux, and soil CH_4 concentration along the P-line. (a) Topography. Base level of specific altitude is located on the surface of the pond. (b) CH_4 fluxes with soil temperature and moisture in June 2001. CH_4 fluxes are indicated in a log scale. P_0-P_{16} : trace $(-8 < F < 8 \mu g C m^{-2} h^{-1})$. (c) CH_4 concentration $(\mu L L^{-1})$ in the soil in June 2001.

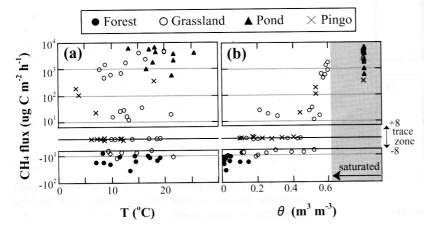


Fig. 5. Relationship between CH₄ flux and (a) soil temperature $(T, ^{\circ}C)$ and (b) soil moisture $(\theta, m^3 m^{-3})$. CH₄ fluxes are indicated in a log scale and CH₄ fluxes beyond the detection limit [-8 < F (μ g C m⁻² h⁻¹) < +8] are depicted in the "trace zone." "Saturated" (gray zone) refers to the zone beyond the detection limit [θ (m³ m⁻³) > 0.60] of a FDR.

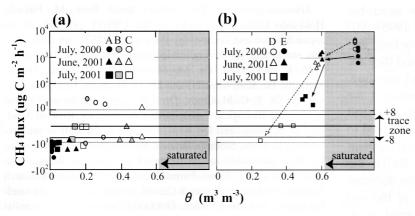


Fig. 6. Relationship between CH₄ flux and soil moisture θ (m³ m⁻³) obtained in the FG-line. (a) Relatively drier plots (A, B, and C). (b) Relatively wetter plots (D and E). CH₄ fluxes are indicated in a log scale and CH₄ fluxes beyond the detection limit [-8 < F (µg C m⁻² h⁻¹) < +8] are depicted in the "trace zone." "Saturated" (gray zone) refers to thezone beyond the detection limit $[\theta \text{ (m³ m⁻³)} > 0.60]$ of a FDR. Arrows in (b) indicate seasonal changes, in CH₄ flux at D (dotted line) and E (solid line).

Table. 2. CH_4 flux from high-latitude ecosystems (>60° N).

La	ti.	Longi.	Location	Soil	Vegetation	Study period	CH ₄ flux range ^a	Reference
Borea	l fore	est	C. 7 - 197				371318(31)	The second second
N	64	W 148	Alaska, US	Silt loam	Spruce	JunSep.	−25 to −11	Gulledge and Schimel 2000
N	62	E 130	Nelegal, Russia	Loamy soil	Larch	JunJul.	-17 to -13	This study
N	60	E 18	Norrliden, Sweden	Orthic podzol	Pine	Annual	-17 to -5	Klemedtsson and
				en species				Klemedtsson 1997
Grass	land							
N	69	E 161	Chersky, Russia	Silts and clays	Festuca spp.	JulAug.	-130-47	Nakano et al. 2000
N	62	E 130	Nelegal, Russia	Loamy soil	Elytriga repens	JunJul.	-11-28	This study
Wetla	nd			weden	2			
N	69	E 161	Chersky, Russia	Floodplain	Horsetail, Carex	Jul.	2.3 - 5.2	Tsuyuzaki et al. 2001
N	67	-N 77	provincia i kalendarje. Nasova i sakada kalendarje.	Wet Tundra		JulAug.	1.5 ± 2.2	Christensen et al. 1995
>N	60		Alaska, UŞ	Wet Tundra	4 -27)10	Warm season	0.3 ± 0.1	Bartlett et al. 1992
N	62	E 130	Nelegal, Russia	Edge of the	Sanati — contil	JunJul.	2.2-4.8	This study
			s from wetherd-	pond				D. Ir. in blunchas
N	62	E 130	Nelegal, Russia	Pond	<u>-</u>	JunJul.	0.6-2.4	This study
N	60	W 161	Alaska, US	Pond	1 x 03 —	Jul.	0.1 - 2.4	Bartlett et al. 1992

^a Unit: μ g C m⁻² h⁻² for boreal forest and glassland, $\times 10^3$ μ g C m⁻² h⁻¹ for wetland.

(maximum emission: $4.8 \times 10^3 \,\mu g \, C \, m^{-2} \, h^{-1}$) in July 2000, whereas in 2001 CH₄ uptake occurred at this site (maximum uptake: $-8 \,\mu g \, C \, m^{-2} \, h^{-1}$).

Comparison with previous studies

Due to the lack of CH₄ flux data in Siberia, the values measured in Alaska and northern Europe have been representative of the Boreal climatic zone for the estimation of global CH₄ emission (Bartlett and Harriss 1993; Potter et al. 1996).

The values of the CH₄ fluxes which were observed in forests, edge of the pond, and pond were similar to those previously reported -25 to -5 μg C m⁻² h⁻¹ for forests and $0.1-5.2 \times 10^3$ μg C m⁻² h⁻¹ for wetlands in Alaska and northern Europe (Table 2).

DISCUSSION

Thermokarsts in Siberia develop due to subsidence

associated with melting permafrost after the occurrence of severe forest fires (Desyatkin 1993). At the initial stage of Alas formation, a pond is formed in the Alas; the pond then gradually dries up due to the small amount of precipitation. As the pond dries up, heavy accumulation of salts takes place in the Alas soil (Desyatkin 1993).

We ascribed the increase in the soil moisture in the Alas grassland to the thermokarst geomorphology. The increase in EC occurred due to salt accumulation and the low precipitation; the increase of the pH and organic carbon content was associated with the forest fires that supplied ash and organic matter to the soil (Sawamoto et al. 2000).

The CH_4 dynamics in the Alas, as determined in this study, showed that, although no detectable CH_4 flux occurred on the salt-accumulated Pingo (Fig. 4), overall the Alas acted as a source of CH_4 emission (Fig. 2).

CH₄ flux from soil occurs as a net CH₄ production and consumption in soil, because an aerobic zone develops

at the surface of the soil aggregates and an anaerobic zone develops inside the aggregates (Conrad 1995). Microbiological processes regulate CH4 production and consumption in soil (Mer and Roger 2001), and they are strongly affected by the soil temperature and moisture. Dunfield et al. (1993) reported that the increase of the soil temperature promoted both CH4 production and consumption in soil. Increase in the soil moisture led to the increase of CH₄ production due to the development of anaerobic conditions (Mer and Roger 2001); the increase in the soil moisture also resulted in the decrease of CH₄ consumption due to the depression of the soil gas diffusivity (Castro et al. 1995; MacDonald et al. 1997; Hu et al. 2001). Our study did not reveal any relationship between the CH₄ flux and soil temperature (Fig. 5), presumably due to the positive temperature dependency of both CH₄ production and consumption in the soil. The forest soil with a low moisture showed a steady CH₄ uptake, while a large CH₄ emission was observed near and in the pond (Figs. 2 and 5). These facts clearly indicate that CH₄ production and consumption strongly depended on the soil moisture.

On the other hand, soils at the top of the Pingo did not show any detectable CH₄ fluxes, in spite of the low soil moisture content which was similar to that of the forest soil (Figs. 2 and 4), presumably due to the higher EC values in the Pingo soil than in the forest soil (Table 1). Nesbit and Breitenbeck (1992) reported that KCl (3.5 cmol_c kg⁻¹) inhibited CH₄ oxidation in soil. In addition, MacDonald et al. (1997) showed that sodium inhibited CH₄ oxidation in soil. In this study, the EC was very high (141 mS m⁻¹) at a 30 cm depth in the Pingo soil. EC is an indicator of total cation concentration, and an EC of 141 mS m⁻¹ corresponds to a total cation concentration of 14 cmol₆ kg⁻¹ (Kamewada 1991). Furthermore, salt-tolerant vegetation, such as Puccinellia tenuiflora and Artemisa commutata, was dominant at the top of the Pingo. However, although the pH on the surface of the Pingo soil was about 6—favorable conditions for the growth of methanotrophs (Hütsch et al. 1994)-CH, uptake was not observed. Therefore, salinity rather than pH may influence the CH₄ flux. Christensen et al. (1995) reported that CH₄ production increased with increasing organic carbon content, and Dunfield et al. (1993) showed that both CH₄ production and consumption were highest at the top of mineral soils. Therefore, a higher CH₄ production may occur in the grassland compared with the forest.

These results suggest that Alas formation due to with forest disturbance increases CH₄ production and decreases CH₄ consumption due to the accumulation of salts and the increase of the soil moisture and organic carbon content in the Alas soil.

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